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RF Project 691 Report No. 31

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REPORT

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THE OHIO STATE UNIVERSITY RESEARCH FOUNDATION

 1314 Kinnear Road Columbus 12, Ohio

To:

DEPARTMENT OF THE NAVY
Office of Naval Research
Contract No. Nonr 495(12)
ARPA Order No. 23-61; Task 1, Item 4

On:

STRUCTURAL AND THERMODYNAMIC PROPERTIES OF POLYATOMIC MOLECULES AT ELEVATED TEMPERATURES

* For the period:

1 January 1964 - 31 March 1964

Submitted by:

David White

Department of Chemistry

Date: 50 September 1964



STRUCTURAL AND THERMODYNAMIC PROPERTIES OF POLYATOMIC MOLECULES AT ELEVATED TEMPERATURES

During this report period a quantitative mass spectrometric investigation of the effect of oxygen pressure on the concentration of gaseous sodium and potassium oxides in equilibrium with solid Na20 and K20 was completed. These data confirm the results of the earlier preliminary studies on the composition of the vapor state. It is still not possible to report new values for the heats of formation of gaseous Na20, Na0, K20, and K0 because of the uncertainty in extimating cross sections. An attempt is currently being made to determine these values experimentally.

The mass spectrometric studies also revealed the existence of both Na_2O_2 and K_2O_2 on the vapor phase. These species are considerably more abundant, relative to the monoxides, than the corresponding species observed in the lithium oxide system. Preliminary matrix isolation studies in the infrared region suggest however that these are peroxides rather than the rhombic form assigned to Li_2O_2 .

Infrared studies of matrix-isolated $\text{MgF}_2(g)$ were initiated during the report period. The present data are not sufficiently complete to warrant a discussion of the molecule structure.

The manuscript describing the investigation of the infrared spectrum of matrix-isolated Al₀O has been submitted for publication.

Investigator_	1	stan	Date	1-/12/64
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•	For The Ohio State I	University Resea	arch Fou	indation •
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